The background of the slide is a microscopic image of sea ice, showing a complex, porous structure with numerous small, circular and irregular voids or bubbles. The color is a mix of light beige and grey. A dark grey rectangular box with a thin white border is positioned in the upper right quadrant, containing the title text in white. The text is centered within the box and reads: "Some ideas on the representation of thermodynamics in sea ice models".

Some ideas on the representation
of thermodynamics
in sea ice models

Martin Vancoppenolle
LOCEAN-IPSL, Paris France

New framework for sea ice phase composition

**TEOS-10
COMPLIANT !**

**NOW
INCLUDING
MINERALS**

$$\phi_{br}(S, T) = [1 - f_{sm}(T)] \cdot \frac{S}{S_{br}(T)}$$

**NEW
!!!**

**CHEAP!
EASY!**

JGR Oceans

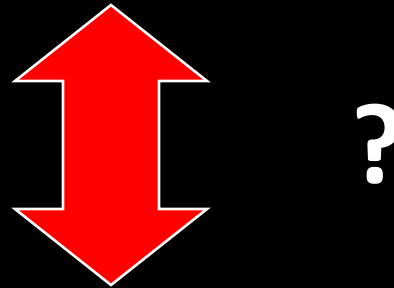
RESEARCH ARTICLE

10.1029/2018JC014611

Thermodynamics of Sea Ice Phase Composition Revisited

Martin Vancoppenolle¹ , Gurvan Madec¹ , Max Thomas², and Trevor J. McDougall³

- Climate model have issues in the sea ice zone
- Sea ice thermodynamics are inadequate in other models than mine
- Need to improve physics to reduce uncertainties



- Climate models are insensitive to uncertainties in sea ice thermodynamics

Overarching issues

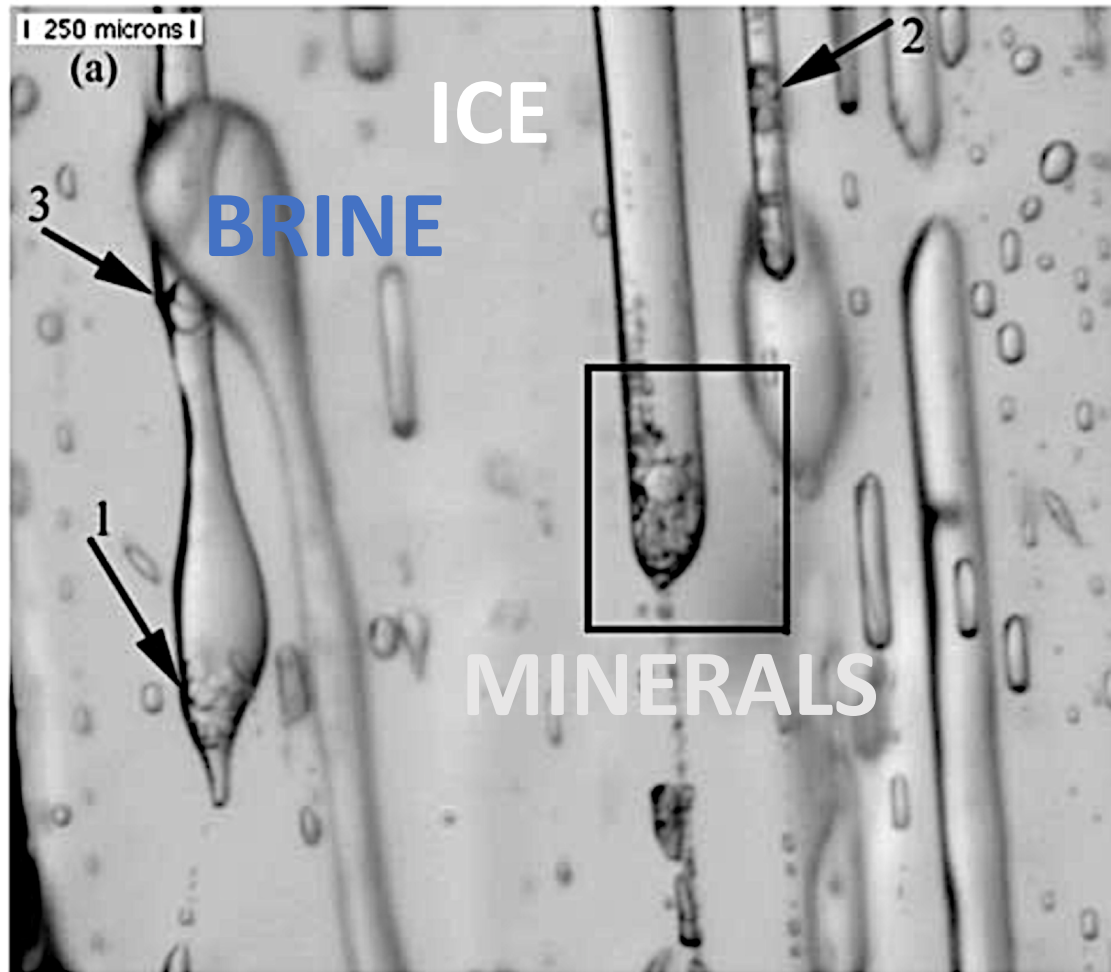
- How well do we know sea ice phase composition?
- For which purposes do we need thermodynamic sea ice model complexity ?
- How should we treat this contradiction in terms of model development ?

Sea ice thermo in CMIP5 models

- o SST = T_f
- o Albedo
- o Conservation of mass and (latent+sensible) heat
- o Ice thickness distribution

Pure ice
Static salinity

Sea ice = ice + brine + minerals



-15°C, natural sea ice thin section (Light et al., 2003)

Key issue : salt

*The chief difference between sea ice and freshwater ice is the presence of salt in the former
(Malmgren, 1927)*

- Composition of sea ice function of T and S
 - H₂O crystals does not like salt
 - $T_{fr} = T_{fr}(S)$

More realistic representation of salt in sea ice thermodynamics ?

- Conservation of mass and (latent+sensible) heat

- Phase composition (ice, brine, minerals)

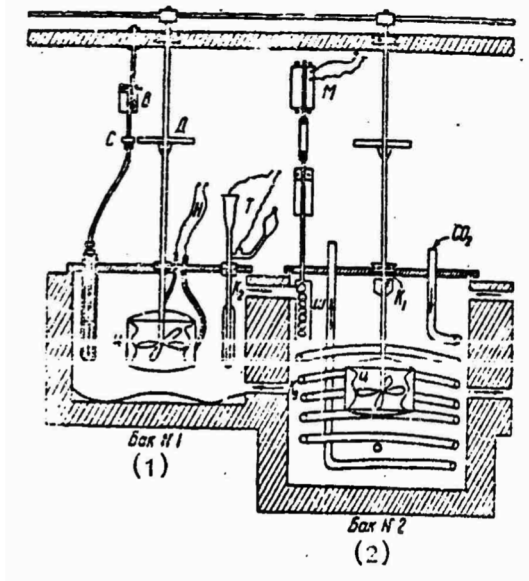
- Enthalpy — $q(S,T)$

- Thermal conductivity — $k(S,T)$

- Salt dynamics — dS/dt

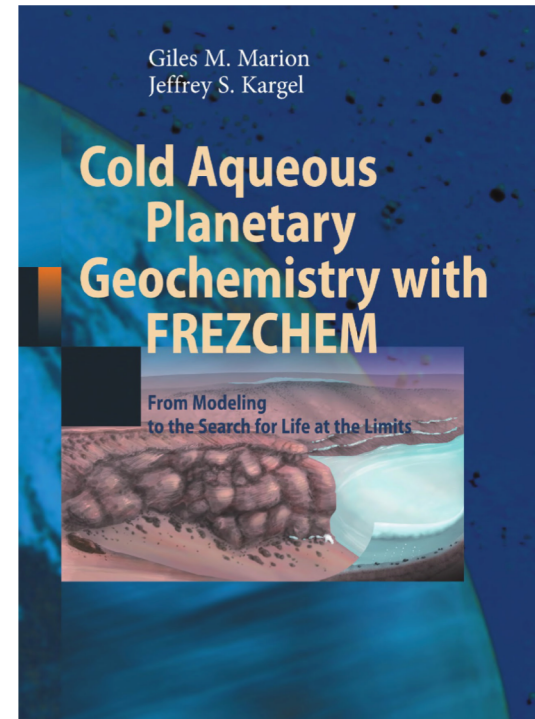
On which base can we develop more realistic sea ice thermodynamics ?

Experimental works



Gitterman 1936

Gibbs-Pitzer theory
for electrolyte solutions



Proposed revised Framework

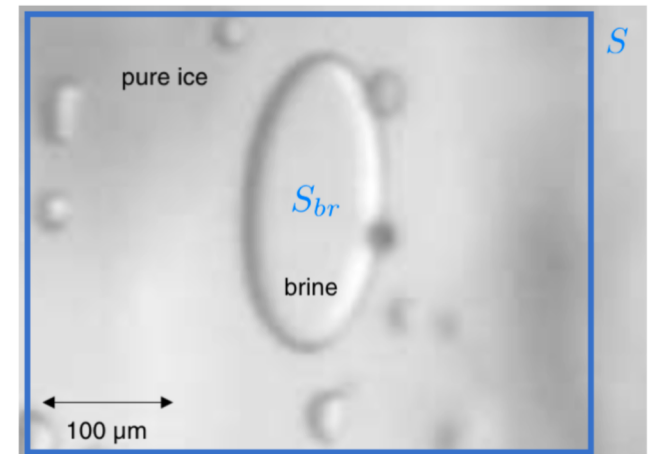
Thermal equilibrium
Standard seawater

$$\phi_{br}(S, T) = [1 - f_{sm}(T)] \cdot \frac{S}{S_{br}(T)}$$

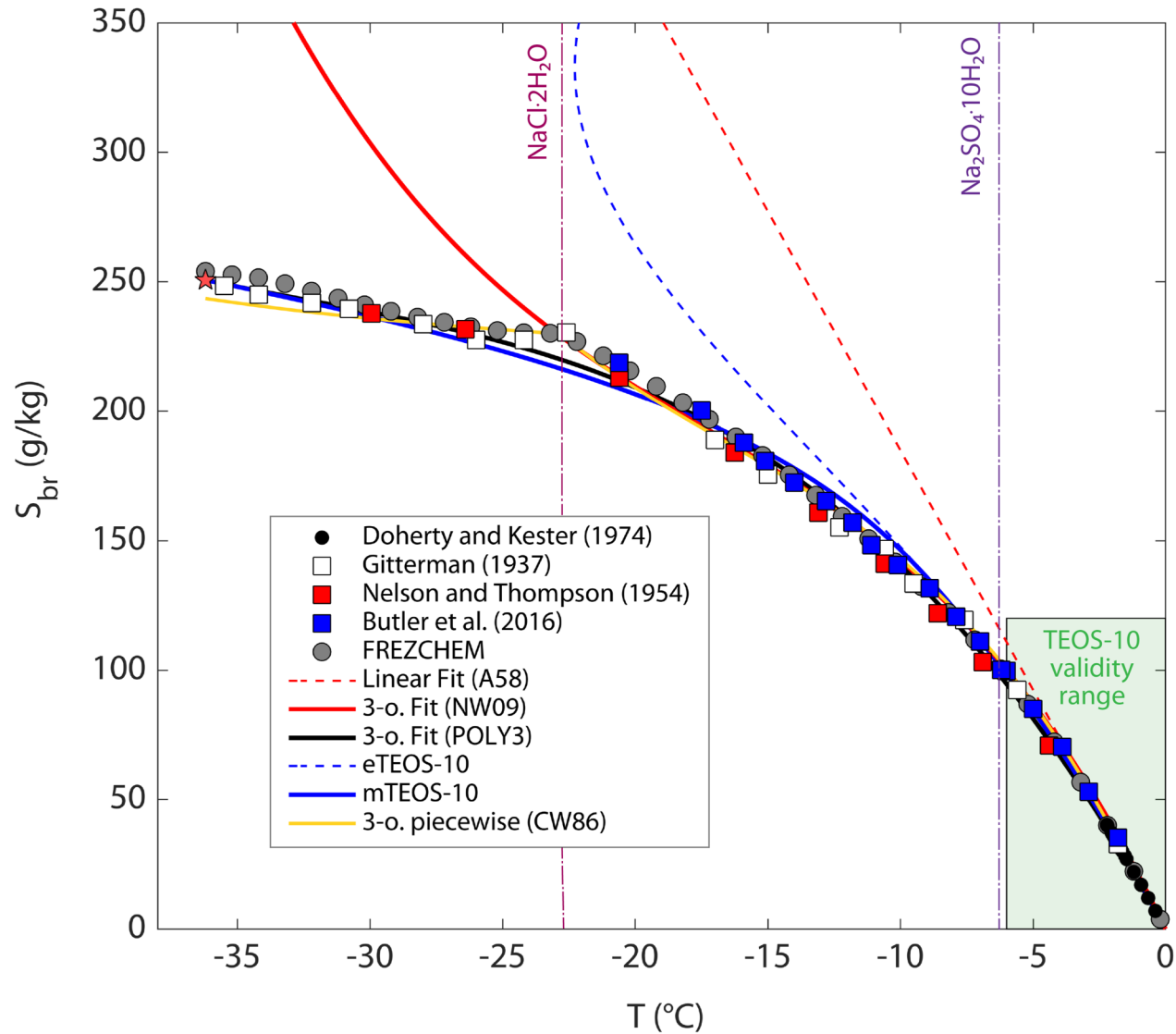
ϕ_{br} Brine mass fraction

f_{sm} Mass fraction of solid salt
in minerals (specified)

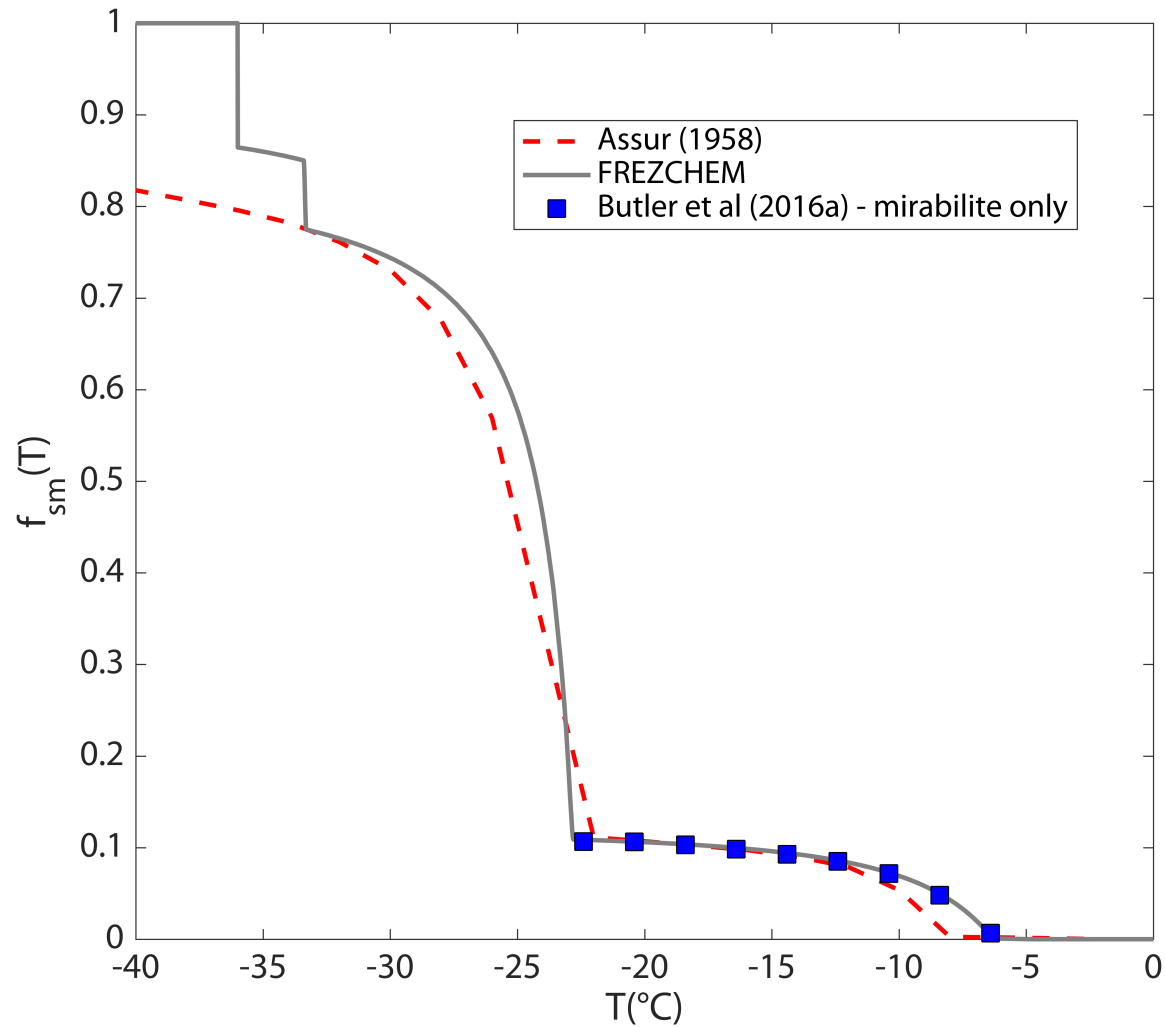
S_{br} Brine salinity (specify)



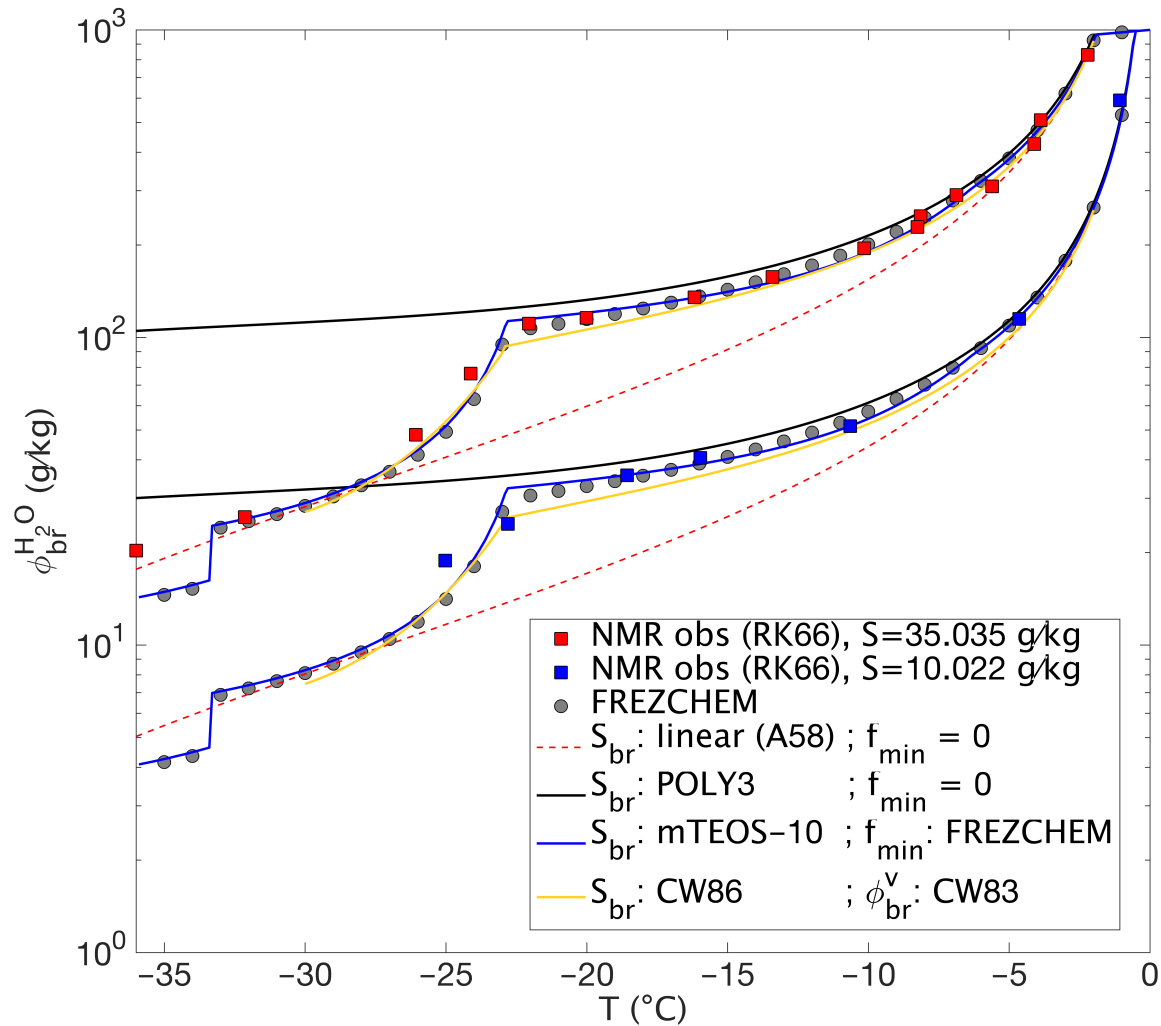
Brine salinity



Minerals



Liquid fraction

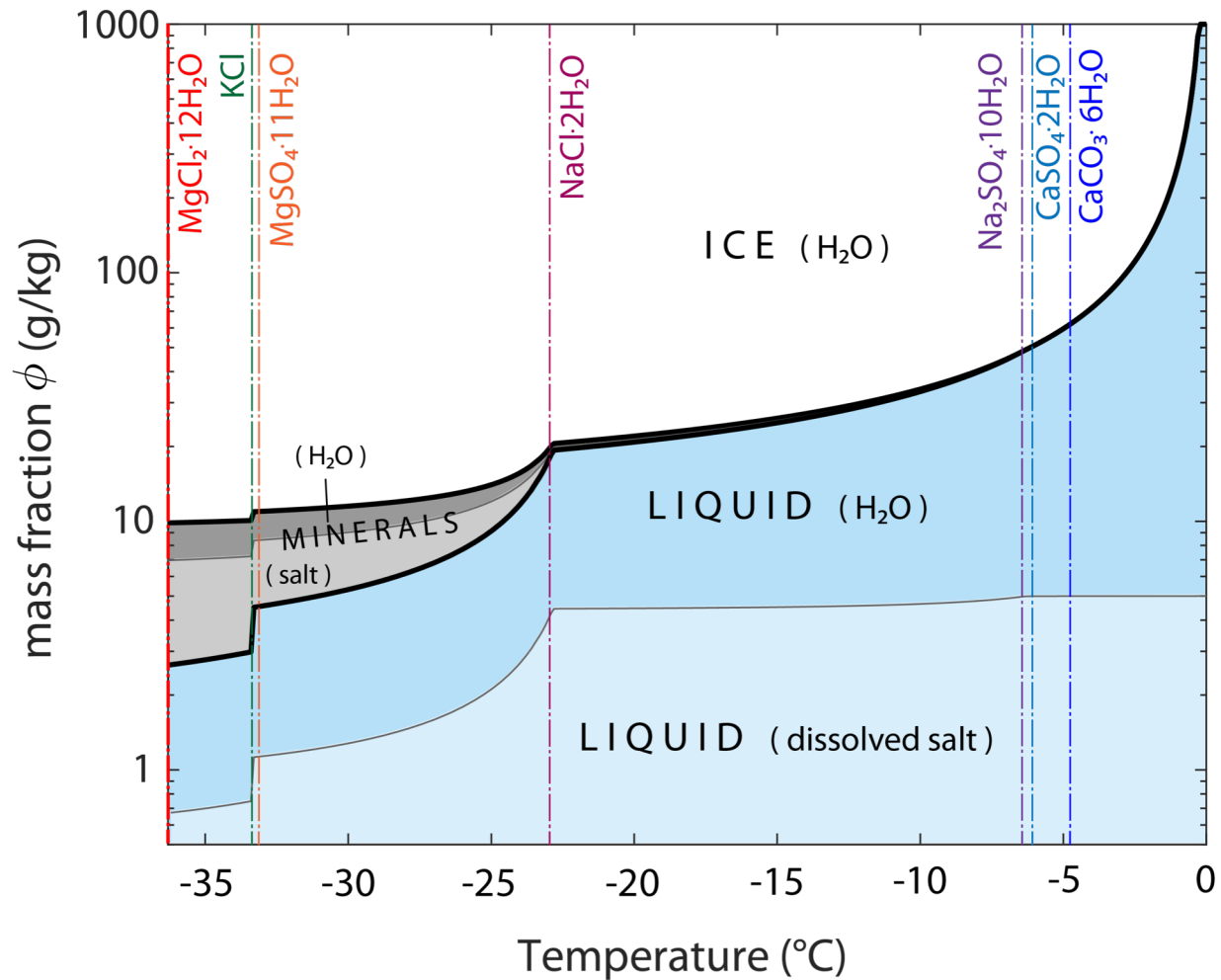


Framework... is appropriate

$$\phi_{br}(S, T) = [1 - f_{sm}(T)] \cdot \frac{S}{S_{br}(T)}$$

- FREZCHEM is the most precise source for all diagnostics
- Uncertainties are reasonable (5-10%)
- One option compatible with TEOS-10

$$\phi_{br}(S, T) = [1 - f_{sm}(T)] \cdot \frac{S}{S_{br}(T)}$$



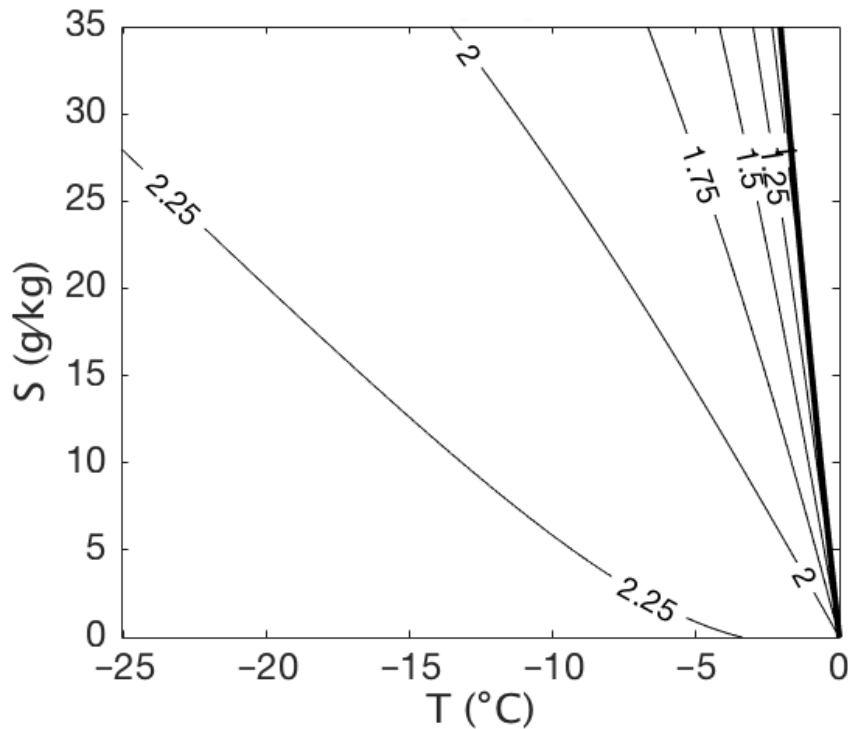
More realistic sea ice thermo ?

- Conservation of mass and (latent+sensible) heat
- Phase composition (ice, brine, minerals)
- Enthalpy — $q(S,T)$
- Thermal conductivity — $k(S,T)$
- Salt dynamics — dS/dt

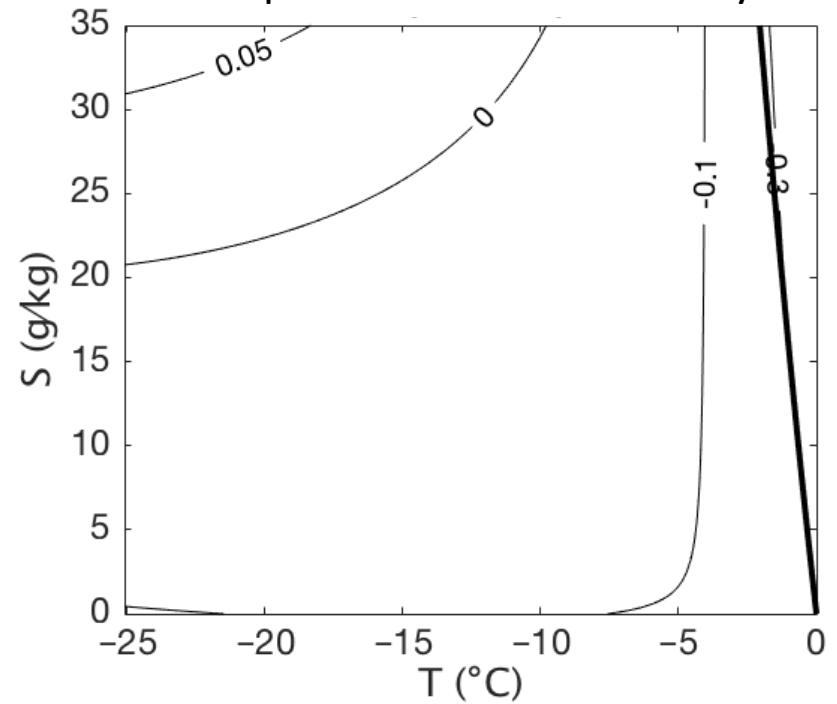
Thermal conductivity

$$k(S, T) = \phi(S, T) \cdot k_{br}[S_{br}(T)] + [1 - \phi(S, T)] \cdot k_i(T)$$

New Theoretical Framework



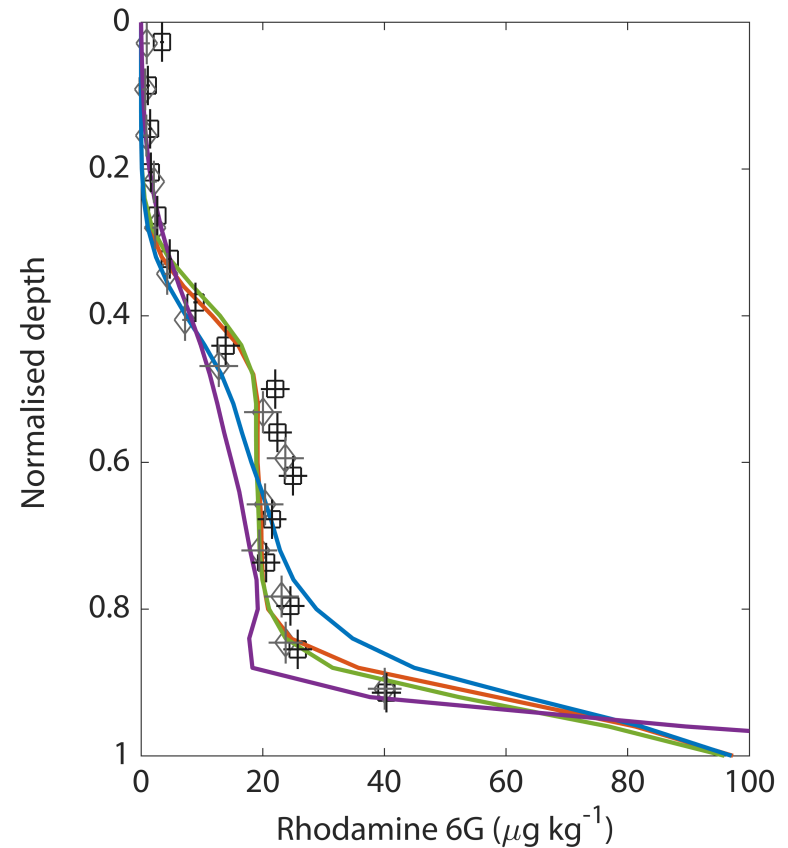
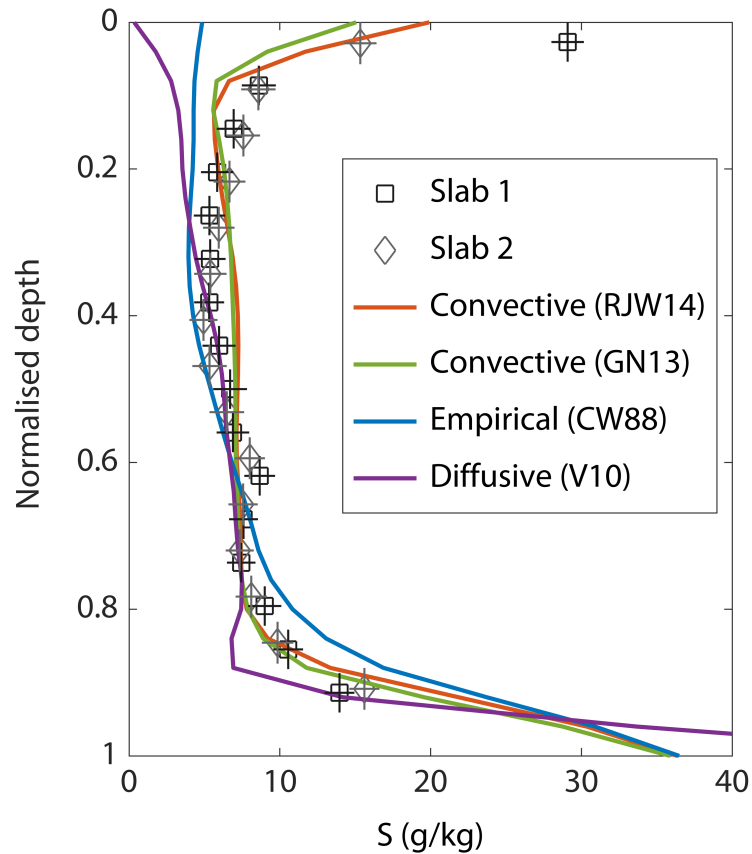
Empirical obs based - theory



Salt dynamics tested in the lab



Salt dynamics tested in the lab



Courtesy Max Thomas

- **How well do we know about sea ice thermodynamics ?**
- When and for which purposes do we need thermodynamic sea ice model complexity ?
- How should we treat this contradiction in terms of model development ?

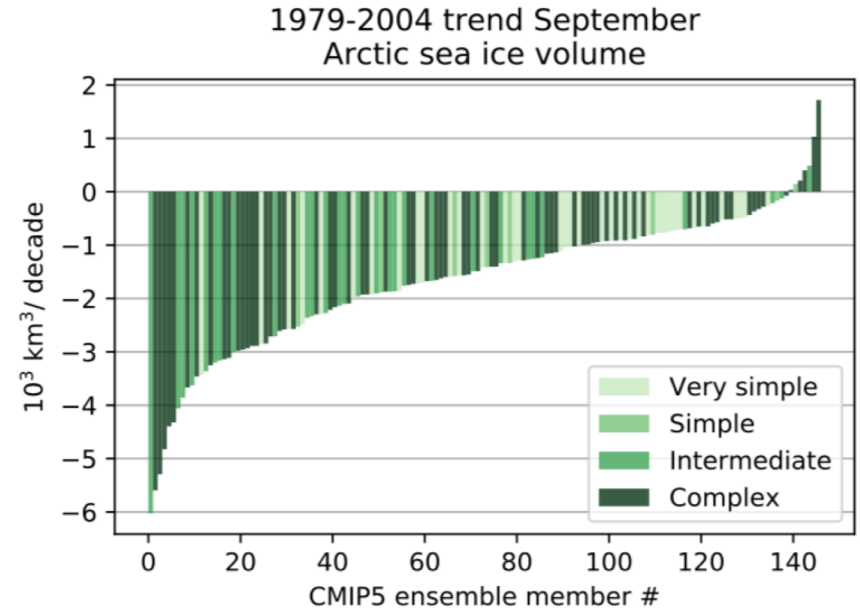
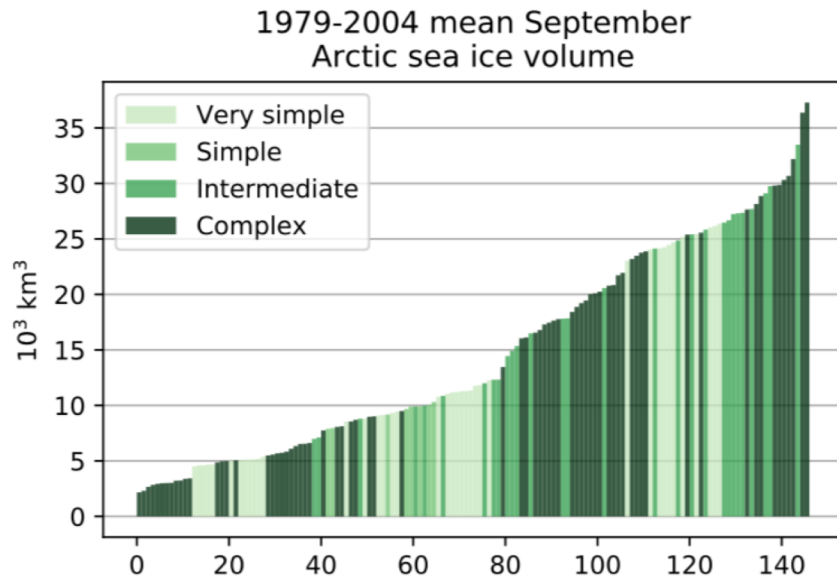
How well do we know about sea ice thermodynamics ?

- o Conservation of mass and (latent+sensible) heat
- o Phase composition (ice, brine, minerals)
- o Enthalpy — $q(S,T)$
- o Thermal conductivity — $k(S,T)$
- o Salt dynamics — dS/dt



- How well do we know sea ice phase composition?
- **When and for which purposes do we need thermodynamic sea ice model complexity ?**
- How should we treat this contradiction in terms of model development ?

Yet climate models seem insensitive to thermodynamic complexity



Courtesy François Massonnet, from Notz et al (in revision)

So what ?

- No need for complicated ice thermo to understand the basics of climate problem
- Benefits from refined thermodynamic representation of sea ice — in terms of impacts
- Modular models are a way to go to choose required complexity
- Remaining uncertainties: ice volume, interactions with atmosphere, and ocean